The Nuclear Magnetic Resonance Spectra of
Some Aconite Bases: Lucaconine and
Lucidusculine

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An alkaloid lucaconine has beeen isolated from Aconitum lucidusculum, Nakai¹⁾. Recently, the alkaloid was identified²⁾ as delcosine³⁾ obtained from Delphinium conolida L. Based on several reactions of delcosine, Marion and others⁴⁾ assumed that the alkaloid possesses the same carbon-nitrogen framework and oxygen pattern as lycoctonine II⁵⁾ and suggested a tentative structure I. However, there has been no direct evidence for this substance bearing the CH₂OCH₃ group for R.

In Aconitum and Delphinium alkaloids, R is generally the methyl, hydroxymethyl or methoxymethyl group⁶. However, the possibility of the presence of the hydroxymethyl group in lucaconine can be excluded, since corresponding aldehydic or acidic products have not been obtained by the oxidation conditions which are expected to lead to these products^{1,4,7}). The observation of spectrum of lucaconine has been made to determine whether R is methyl or methoxymethyl. The NMR spectrum at 40 Mc. of lucaconine in pyridine solution is shown in Fig. 1. In this figure, although the other Aconitum alkaloids in which R is methyl have a strong singlet absorption characteristic of

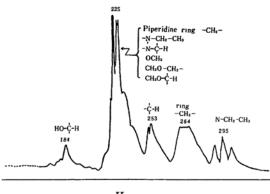
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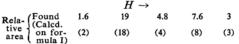
²⁾ T. Amiya and T. Shima, in preparation.

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⁶⁾ O. E. Edwards, L. Marion and K. H. Palmer, Can. J. Chem., 36, 1097 (1958); K. Wiesner, D. L. Simmons and L. R. Fowler, Tetrahedron Letters, No. 18, 1 (1959).

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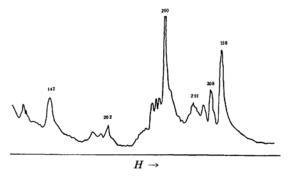
The proton spectrum of lucaconine (taken on Varian 40 Mc. instrument).

-C-CH₃ group at about 318 c.p.s. from the signal of α -proton of pyridine⁸), such absorption can not be recognized for lucaconine. Accordingly, R is concluded to be CH2OCH3 The anticipated singlet peak in lucaconine. due to the methylene group of -C-CH2-OCH3 is, however, obscure⁹⁾ in Fig. 1, probably because of the overlapping with strong peaks of methoxyl groups, and of possible broadening due to the unequivalence of two methylene protons caused by the neighbouring asymmetric carbon atom10). A tentative interpretation based on the suggested structure I is also given in Fig. 1¹¹).

We next examined NMR spectrum of an aconite base lucidusculine12). Recently, one of us (T. A.) proposed formula III for the partial structure of lucidusculine on the basis of some chemical transformations¹³).

III R=CH₂OH, one secondary OH group on the ring. Possible linkage between C_{17} - C_{19} or C_{17} - C_{9} .

IV R=CH3, two secondary OH groups on the ring. Possible linkage between C_{17} - C_{19} or C_{17} - C_{9} .



The proton spectrum of lucidusculine.

The NMR trace at 40 Mc. of lucidusculine in pyridine is shown in Fig. 2. Clearly, the singlet band at 318 c. p. s. is to be ascribed to the methyl group on a quaternary carbon atom. This observation forces one to the conclusion that in lucidusculine R is actually the methyl group. The two hydroxyl groups in lucidusculine may both be secondary, since they are easily acetylated¹². From above described results, a modified formula IV is now suggested as the partial structure of lucidusculine.

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⁸⁾ A. Suzuki, T. Amiya, K. Takahashi and T. Matsumoto, unpublished work.

⁹⁾ In the NMR spectrum of lycoctonine8, the absorption due to methylene group of -C-CH2OH is also ob-

¹⁰⁾ L. M. Jackman, "Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry", Pergamon Press, London (1959), p. 102.

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